

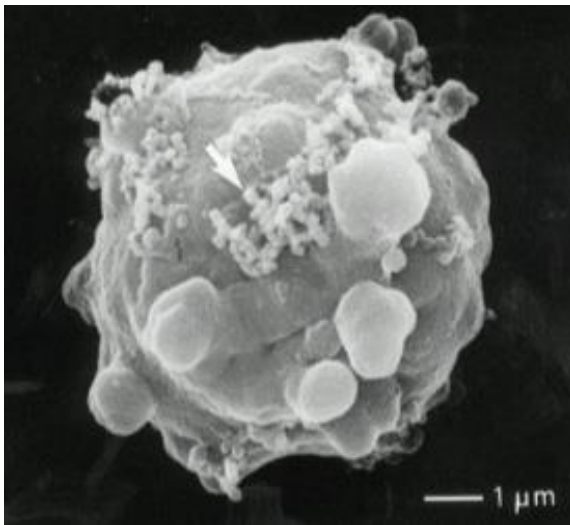
QM/MM study on inhibitor of HIV-1 protease

Graduate School of Science, Kyoto University

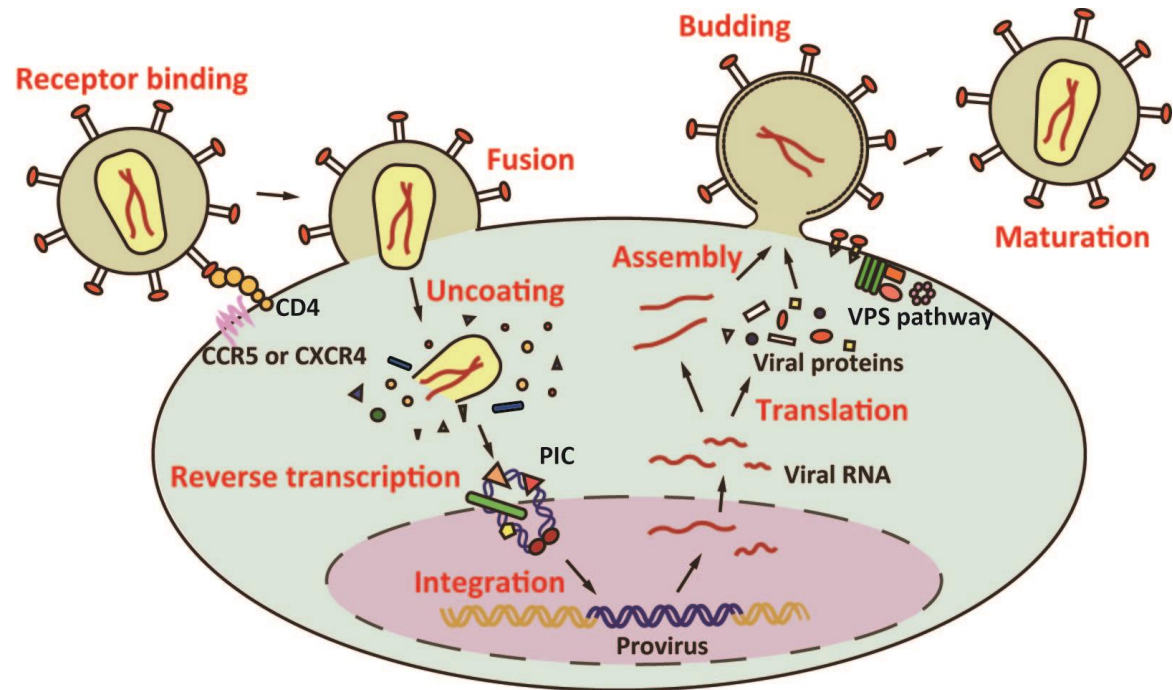
Masahiko Taguchi, Masahiko Kaneso, Shigehiko Hayashi

HIV

- Human Immunodeficiency Virus
- Origin of AIDS



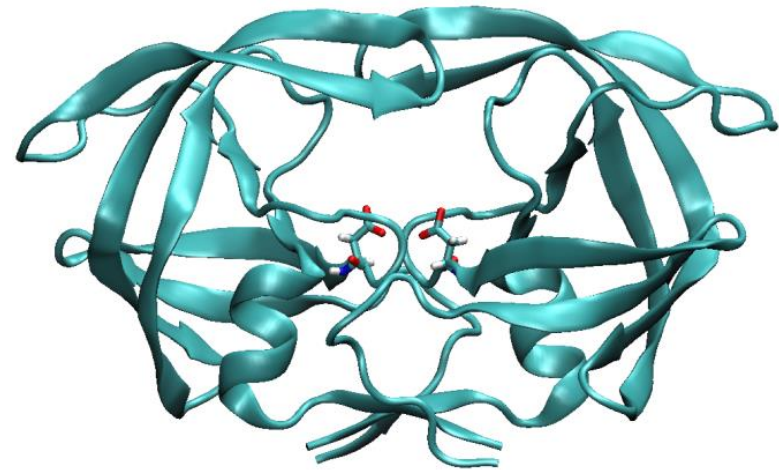
HIV appearing from lymphocytes
Ref: <https://www.pref.hiroshima.lg.jp/soshiki/25/bu-biseibutu2-hivvirus.html>



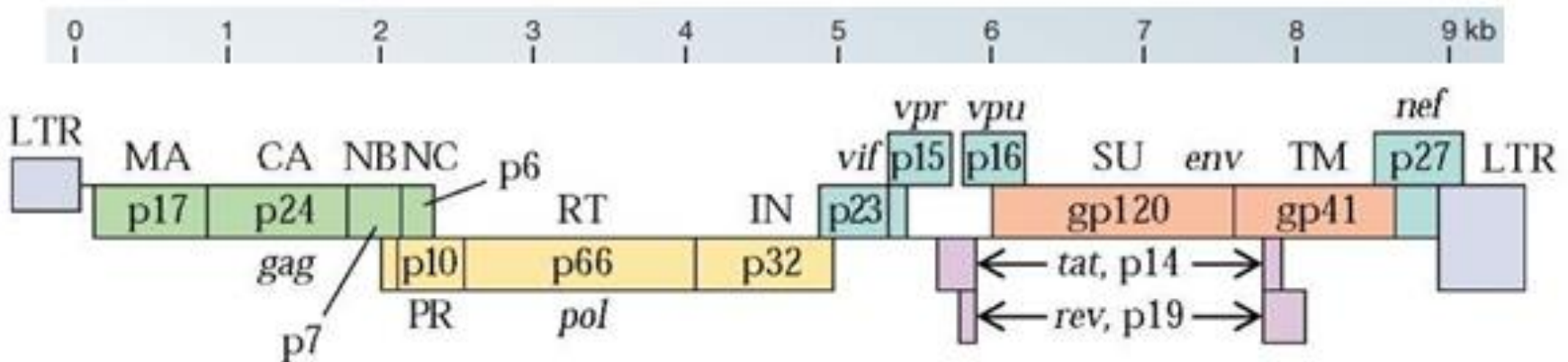
HIV replication in a cell
Ref: <http://colgateimmunology.blogspot.de/2013/12/a-new-mechanism-of-cell-death-by-hiv.html>

HIV-1 protease

- One of protein in HIV virus
- Activated by cutting polyprotein (gag, pol) with hydrolysis



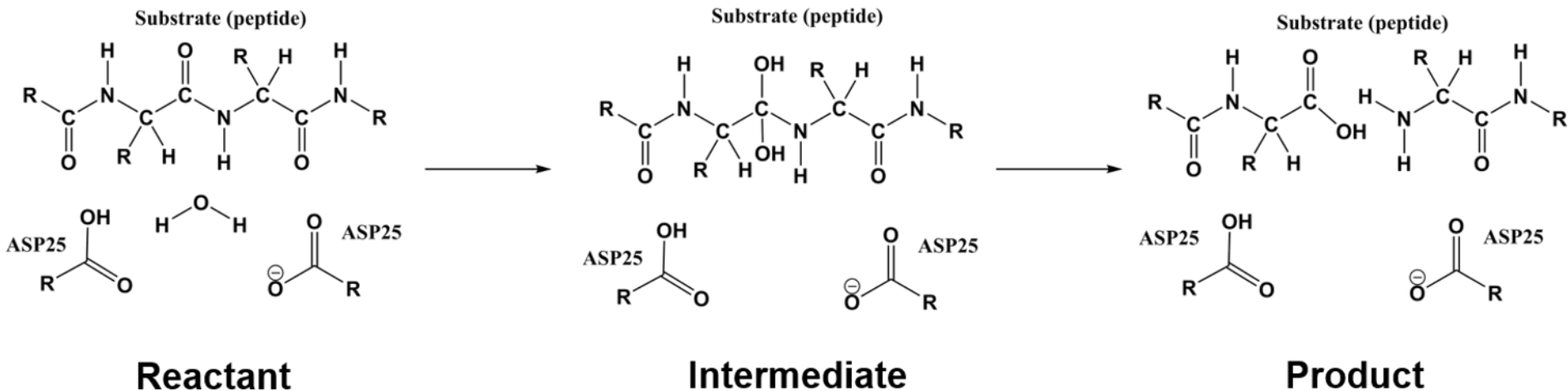
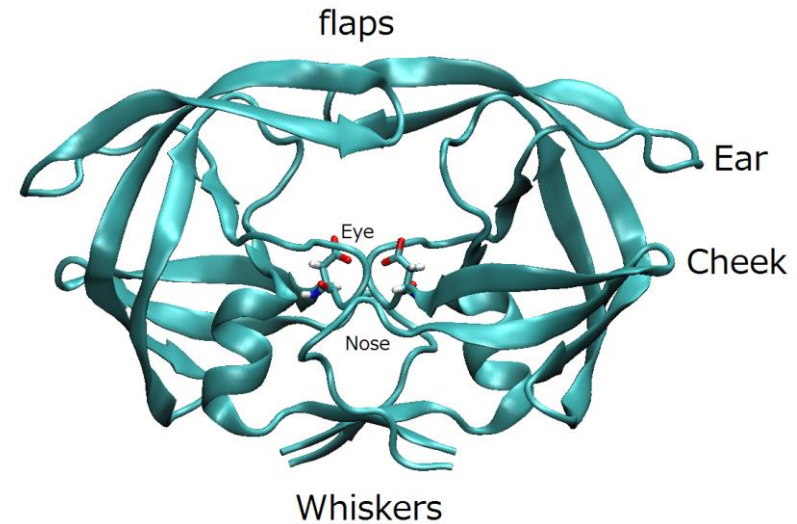
HIV-1 protease



Part of gene of HIV-1 protease

HIV-1 protease

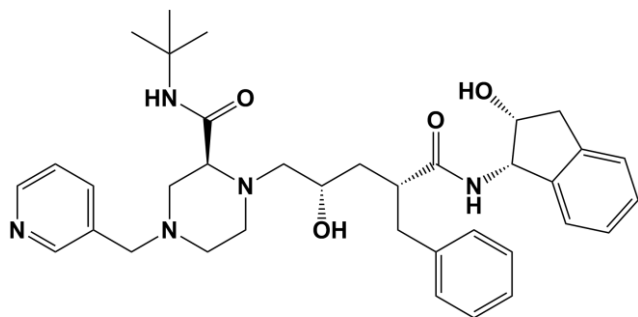
- Homodimer consisted of 99 residues $\times$ 2
- Activation part: two Asp25
- Catalyzing hydrolysis of peptide bonds



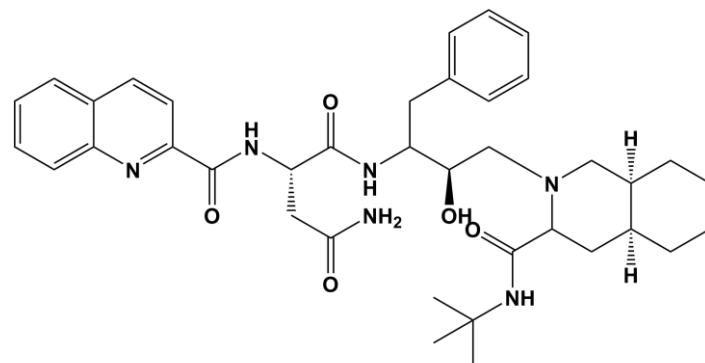
water molecule near chemical reaction part is important!

Inhibitors of HIV-1 protease

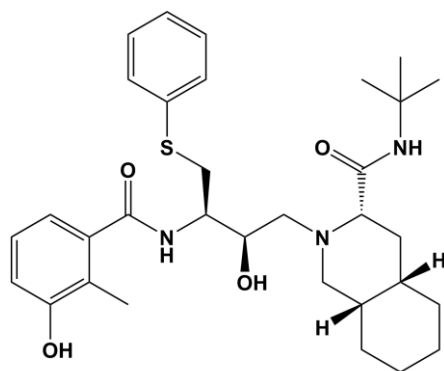
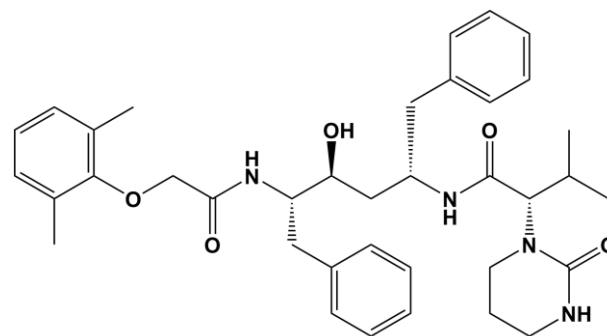
- AIDS medicine
- Transition state(TS) analog molecule
- Serious drug-resistance(high mutation rate)



INDINAVIR

**SAQUINAVIR**

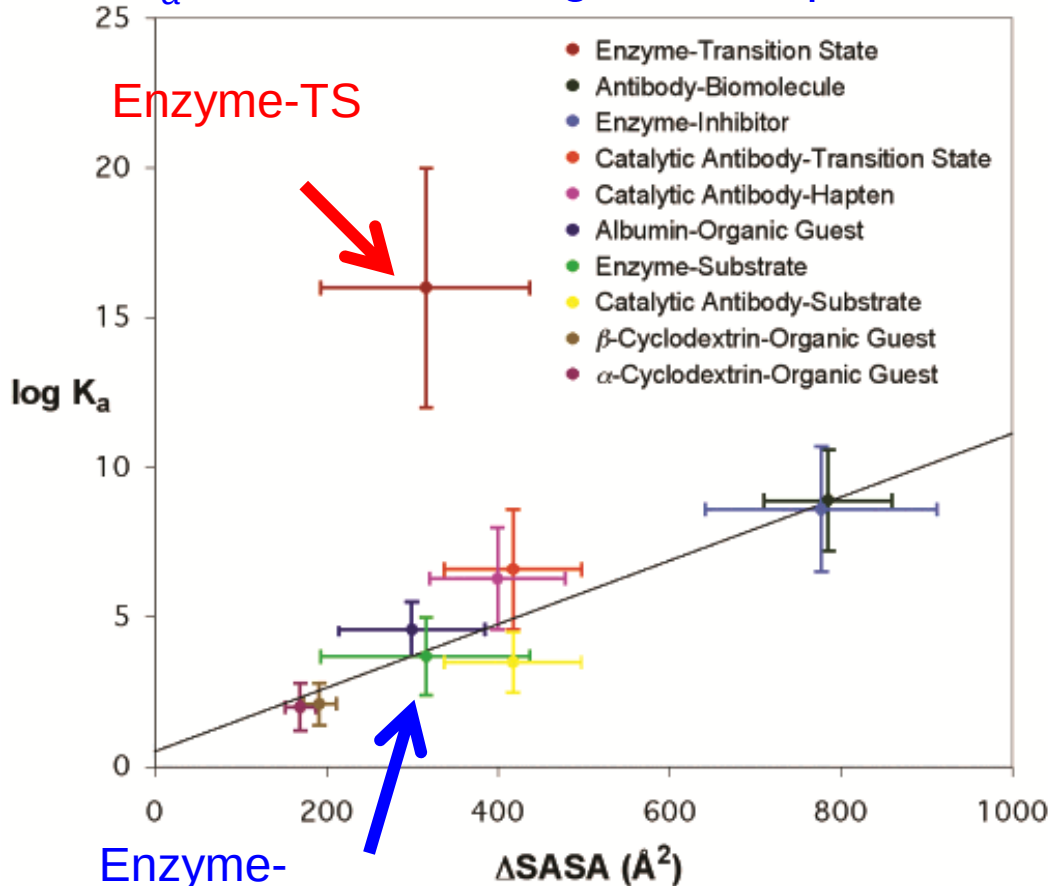
Example of inhibitor of HIV-1 protease

**NELFINAVIR**

LOPINAVIR

Stabilization of TS in enzyme

K_a of various host-guest complexes

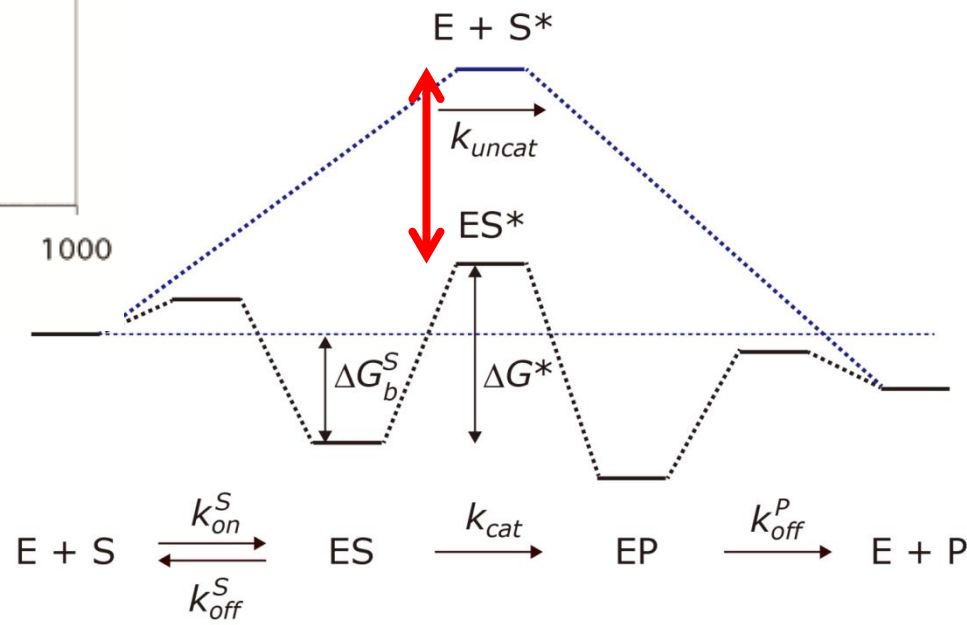


Zhang, Houk, ACR (2005)

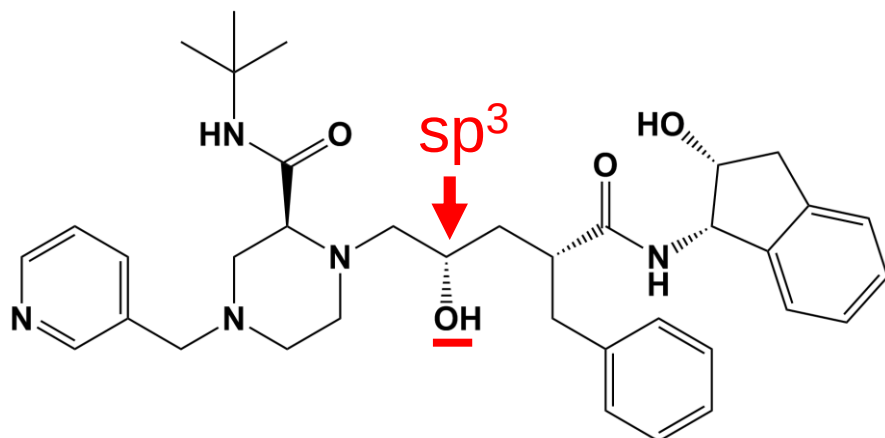
K_a of TS substrate

$$\log K_a \propto -\log \frac{k_{cat}}{K_m k_{uncat}}$$

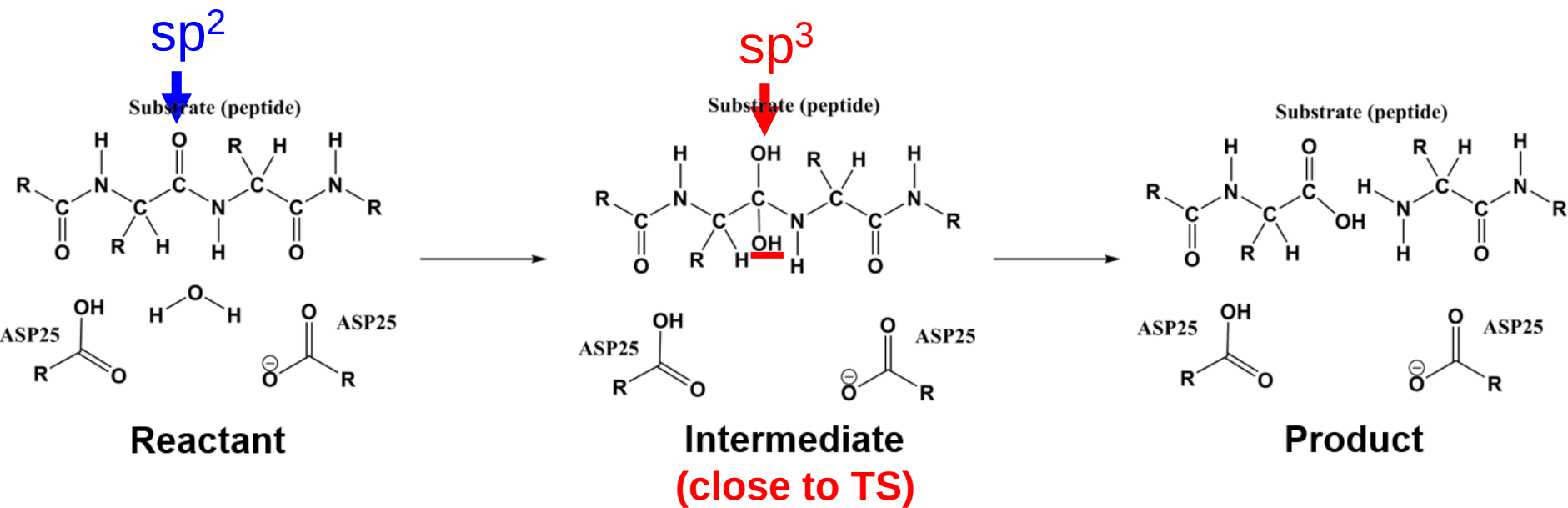
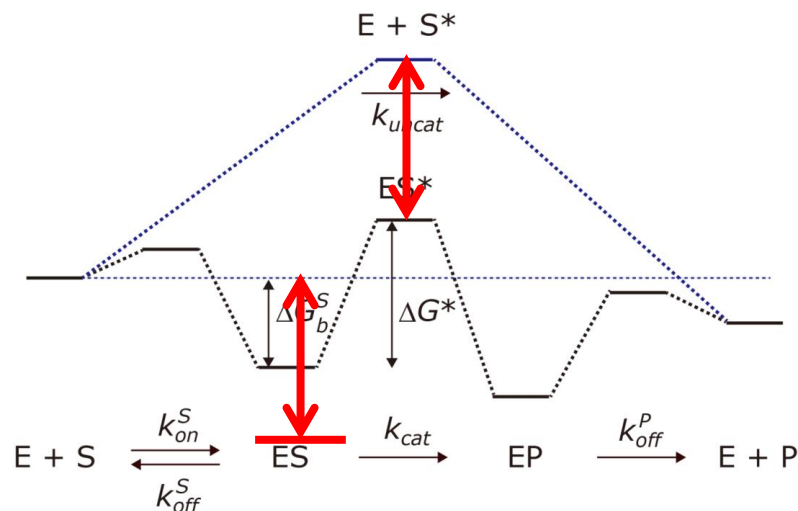
Exceptionally large

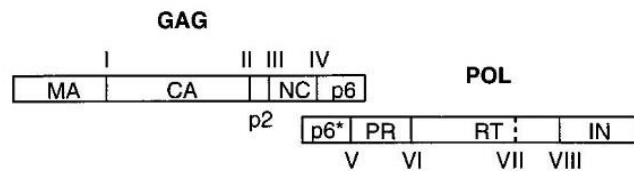


TS analog inhibitor



Indinavir





Drug Resistance

Shock et al., J. Biol. Chem. **271**, 31957 (1996)

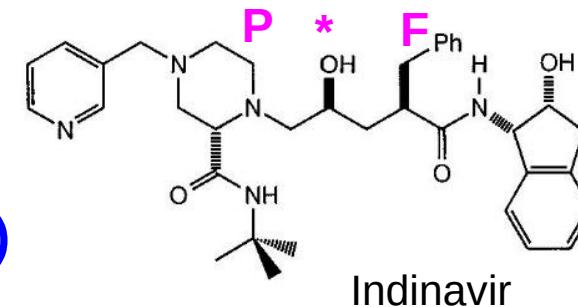
Reaction rate of natural substrate molecule

Cleavage site	Peptide sequence	Wild-type	4X	M46I/L63P	V82T/I84V
		$M^{-1} s^{-1}$	$M^{-1} s^{-1}$	$M^{-1} s^{-1}$	$M^{-1} s^{-1}$
I. MA(p17)/CA(p24)	NQVSQNY*PIVQNI	17,980	940	65,100	830
II. CA/p2	GHKARVL*AEAMSQ	720	310	790	220
III. p2/NC(p7)	TNSATIM*MQRGNF	27,970	3,440	76,300	1,770
IV. NC/p6	KGRPGNF*LQSRPE	370	140	490	43
V. p6*/PR	GTVSFNF*PQVTLW	113,400	6,810	221,400	3,760
VI. PR/RT	IGCTLNF*PISPIE	8,930	430	14,600	300
VII. RT (internal)	IVGAETF*YVDGAA	11,400	227	14,160	130
VIII. RT/IN	AGIRKVL*FLDGID	845	58	1,850	38

Dissociation constant of inhibitor molecule

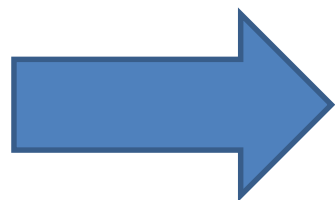
Inhibitor	Wild-type	4X	M46I/L63P	V82T/I84V
	nM	nM	nM	nM
MK-639	0.40	21.6	0.47	23.7
ABT-538	0.19	28.5	0.15	30.1
Ro 31-8959	0.33	11.9	0.37	13.3
VX-478	0.57	5.8	0.53	10.1

1. High activation part is largely effected by its mutation
2. Inhibitor is a mimic of high activation site (e.g. F * P)?
3. Low activation part (determines overall activation) is smally effected by its mutation (e.g. site II)



Purpose

1. Precise identification on transition state of natural substrate molecule
2. Clarifying drug binding affinity and molecular mechanism of enzymatic reactive activity for drug-resistant mutant
3. Design of drug molecule for controlling both drug binding affinity and reactivity of natural substrate



Correlation between chemical reaction part and surrounding protein environment is important

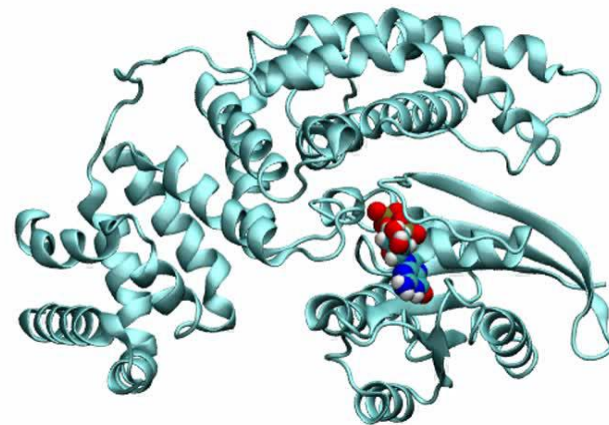
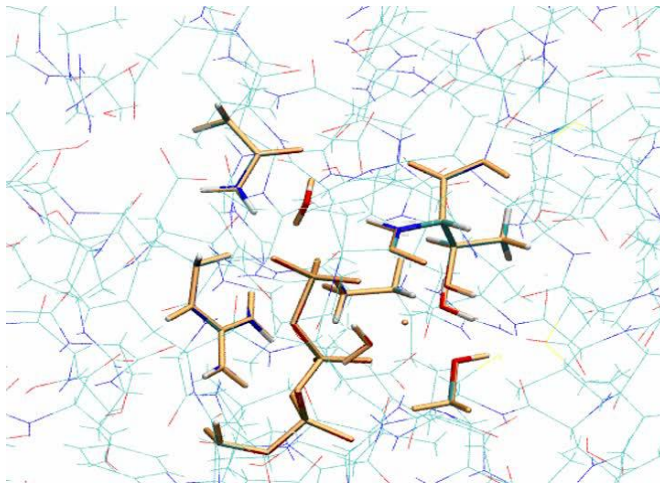
QM/MM RWFE method

Complete separation between

- QM/MM calculation of a chemical reaction and
- MD simulation for protein conformational sampling

QM/MM free energy (FE) geometry optimization

Geometry optimization of the QM molecule on a **free energy surface** defined with the MM conformational thermal distribution obtained by MD simulations



~ 1 μ second (> 1,000 times longer than direct QM/MM-MD)

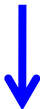
QM/MM-FE Geometry Optimization

Nagaoka and co-workers

Free energy functional

$$F[\Psi] = -k_B T \ln \int \int d\mathbf{R} d\mathbf{X} \exp \left[-\beta E^{QM/MM} (\{\Psi(\mathbf{R}, \mathbf{X})\}, \mathbf{R}, \mathbf{X}) \right]$$

Ψ : electronic wf, $\mathbf{R}$: QM coordinates, $\mathbf{X}$: MM coordinates

Give up sampling of $\mathbf{R}$ 

$$\underline{F[\Psi, \mathbf{R}]} = -k_B T \ln \int d\mathbf{X} \exp \left[-\beta E^{QM/MM} (\{\Psi(\mathbf{R}, \mathbf{X})\}, \mathbf{R}, \mathbf{X}) \right]$$

Free energy surface of QM coordinates

Gradient on FE surface: mean gradient

$$\frac{\partial F}{\partial R_i} = \left\langle \frac{\partial E^{QM/MM}}{\partial R_i} \right\rangle_{\mathbf{x}}$$

Problem: Electronic WF depends on $\mathbf{X}$.

Its computational cost is not very different from a direct QM/MM-MD.

Mean Field (MF) QM/MM Method

Yamamoto

$$F[\Psi, \mathbf{R}] = -k_B T \ln \int d\mathbf{X} \exp \left[-\beta E^{QM/MM} (\{\Psi(\mathbf{R}, \mathbf{X})\}, \mathbf{R}, \mathbf{X}) \right]$$

MF approximation

$$\Psi(\mathbf{R}, \mathbf{X}) \rightarrow \Psi_{MF}(\mathbf{R})$$



MF-QM/MM

$$F[\Psi_{MF}, \mathbf{R}] = -k_B T \ln \int d\mathbf{X} \exp \left[-\beta E^{QM/MM} (\{\Psi_{MF}(\mathbf{R})\}, \mathbf{R}, \mathbf{X}) \right]$$



Variational
solution

$$f^{QM/MM}(\mathbf{d}, \mathbf{R}) = f^0(\mathbf{d}, \mathbf{R}) + \hat{\mathbf{q}}^t(\mathbf{R}) \underline{\bar{\mathbf{V}}^{Cl}(\mathbf{d}, \mathbf{R})}$$

MF of MM region is computed by MD.

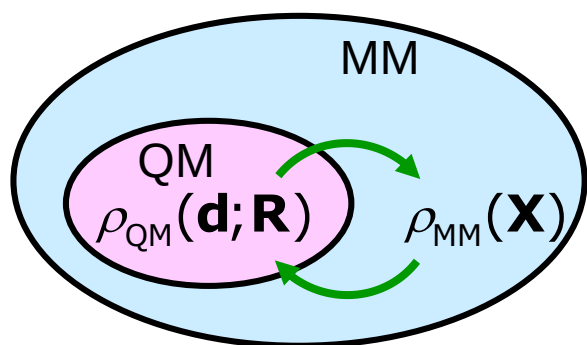
QM/MM RWFE-SCF Method

Kosugi and SH, JCTC (2012)

Kosugi and SH, JACS (2012)

Problem of QM/MM-FE

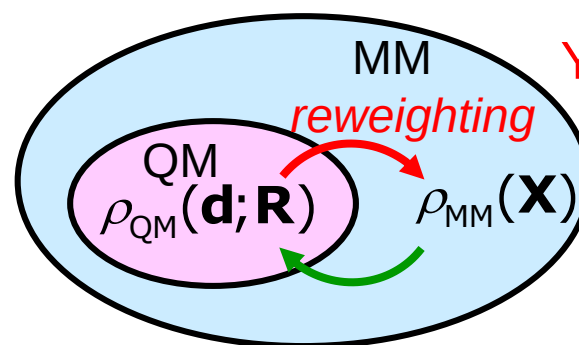
Conventional



Frequent iteration of QM/MM opt. and MD sampling is necessary.

Convergence problem!

Present method



Yang et al.

MM distribution is reweighted.

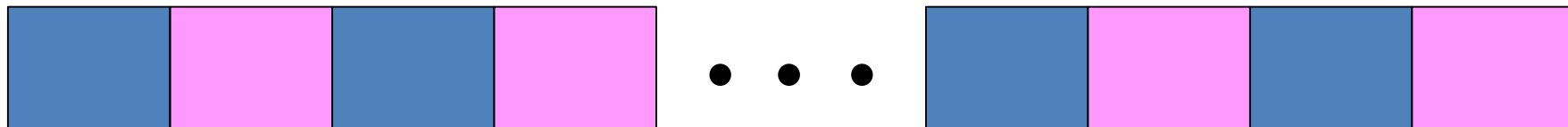
Much quicker convergence of MM thermal distribution

$$\rho_{MM}(\mathbf{d}, \mathbf{R}; \mathbf{X}) = \frac{\exp\left[-\beta \left\{E^{QM-MM}(\mathbf{d}, \mathbf{R}; \mathbf{X}) - E^{QM-MM}(\mathbf{d}_0, \mathbf{R}_0; \mathbf{X})\right\}\right]}{\left\langle \exp\left[-\beta \left\{E^{QM-MM}(\mathbf{d}, \mathbf{R}; \mathbf{X}) - E^{QM-MM}(\mathbf{d}_0, \mathbf{R}_0; \mathbf{X})\right\}\right] \right\rangle_0} \times \rho_{MM}(\mathbf{d}_0, \mathbf{R}_0; \mathbf{X})$$

Well-defined variational method

Procedure

MD QM/MM MD QM/MM ... MD QM/MM MD QM/MM



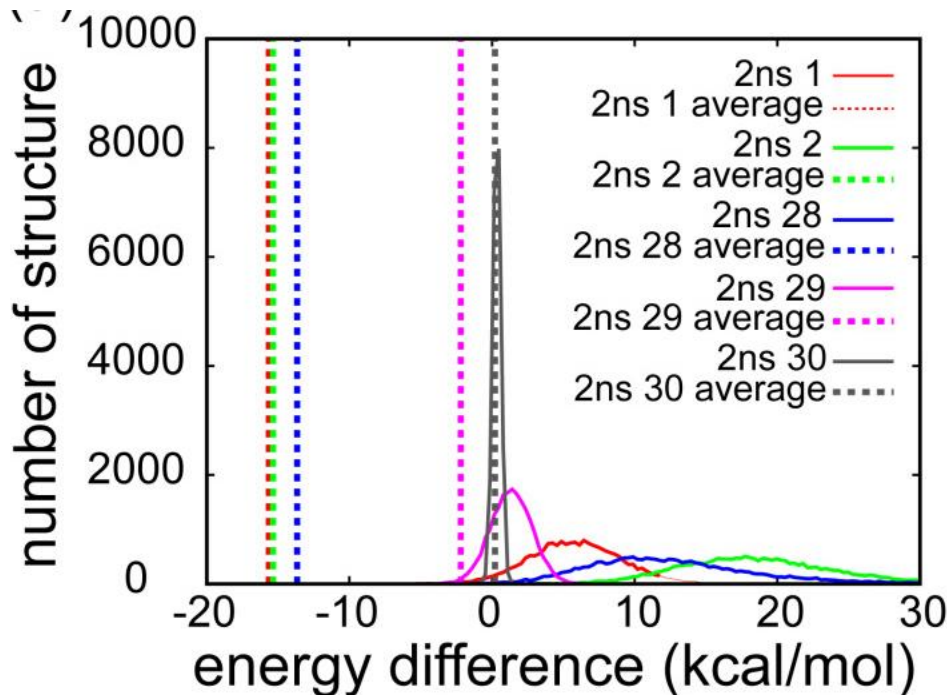
$$\mathbf{R}_{opt}^i \rightarrow \mathbf{R}_0^{i+1}$$

$$\mathbf{q}_{opt}^i \rightarrow \mathbf{q}_0^{i+1}$$

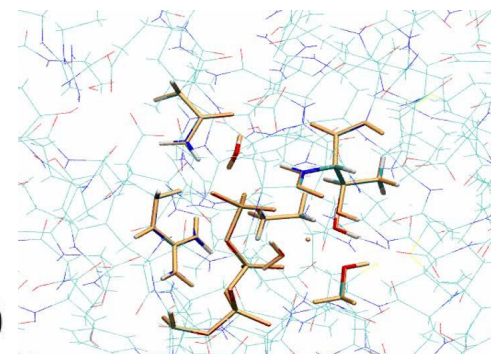
Update

MD

3 – 15 ns
~20,000 confs.



Distribution of energy difference

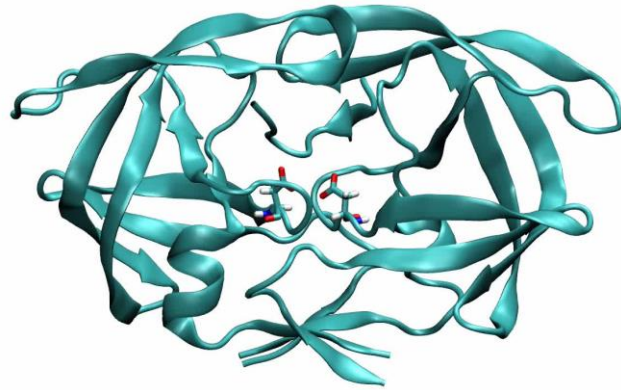


$$\underline{\rho_{MM}(\mathbf{d}, \mathbf{R}; \mathbf{X})} = \frac{\exp\left[-\beta \left\{E^{QM-MM}(\mathbf{d}, \mathbf{R}; \mathbf{X}) - E^{QM-MM}(\mathbf{d}_0, \mathbf{R}_0; \mathbf{X})\right\}\right]}{\left\langle \exp\left[-\beta \left\{E^{QM-MM}(\mathbf{d}, \mathbf{R}; \mathbf{X}) - E^{QM-MM}(\mathbf{d}_0, \mathbf{R}_0; \mathbf{X})\right\}\right] \right\rangle_0} \times \rho_{MM}(\mathbf{d}_0, \mathbf{R}_0; \mathbf{X})$$

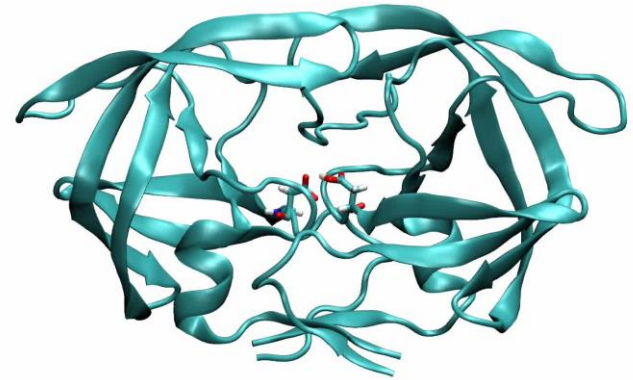
Determination of protonated state

(classical MD)

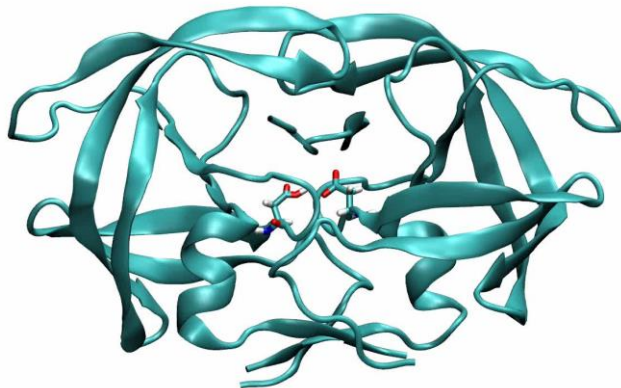
We cannot determine protonated state by X-ray structural analysis!



non-protonated: 300 ns



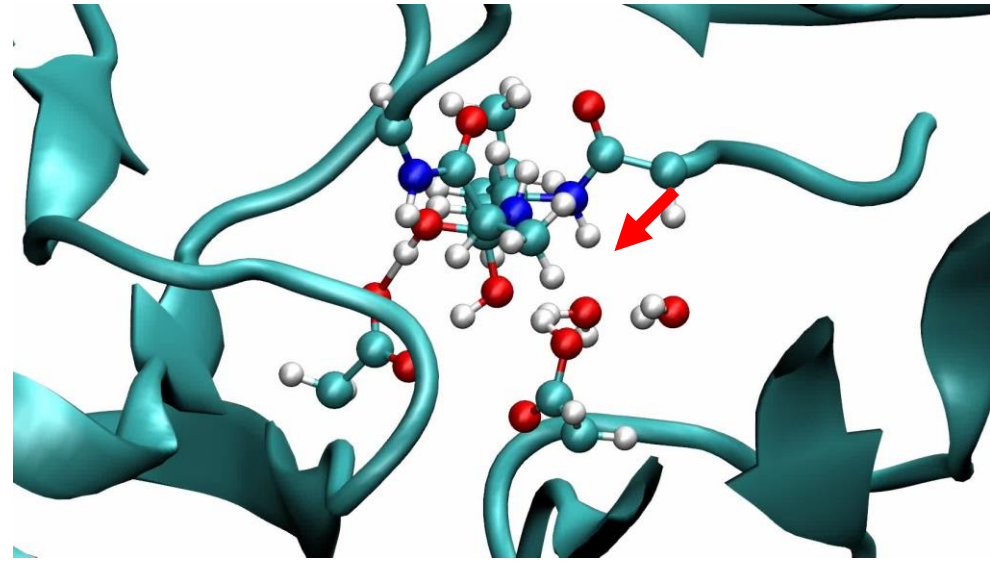
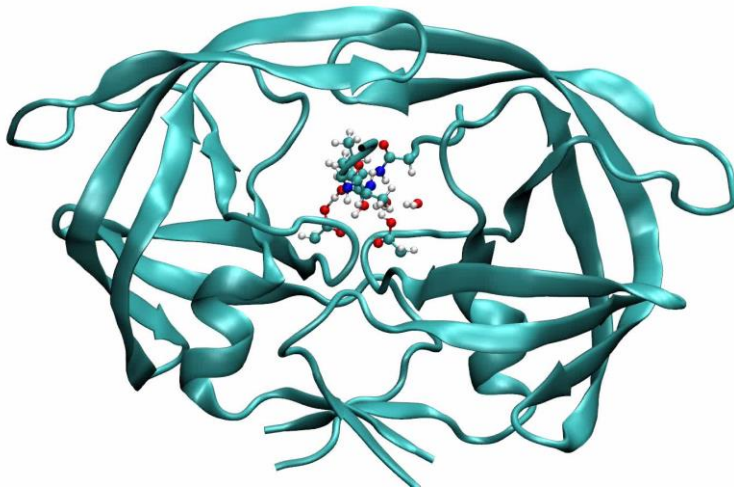
mono-protonated(A): 700 ns



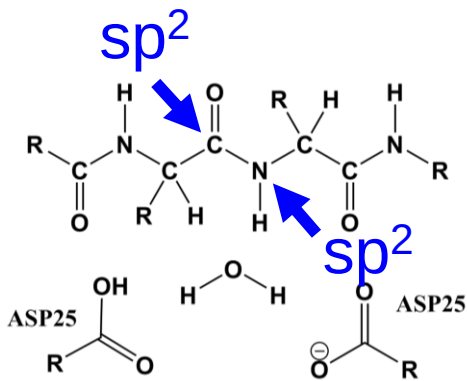
mono-protonated(B): 1000 ns

- non-protonated and mono-protonated (A): substrate dissociated
- mono-protonated(B): stable under 1 μ s MD and water molecules entered activation part

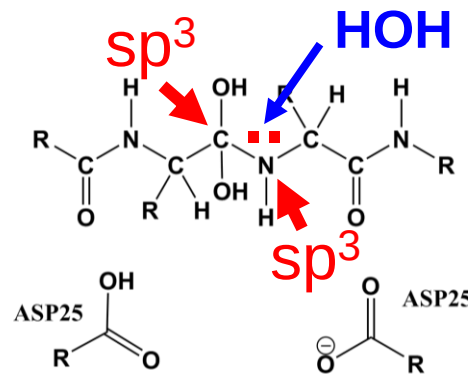
Intermediate state (QM/MM)



Not yet converged ($\sim 6 \mu\text{s}$)



Reactant



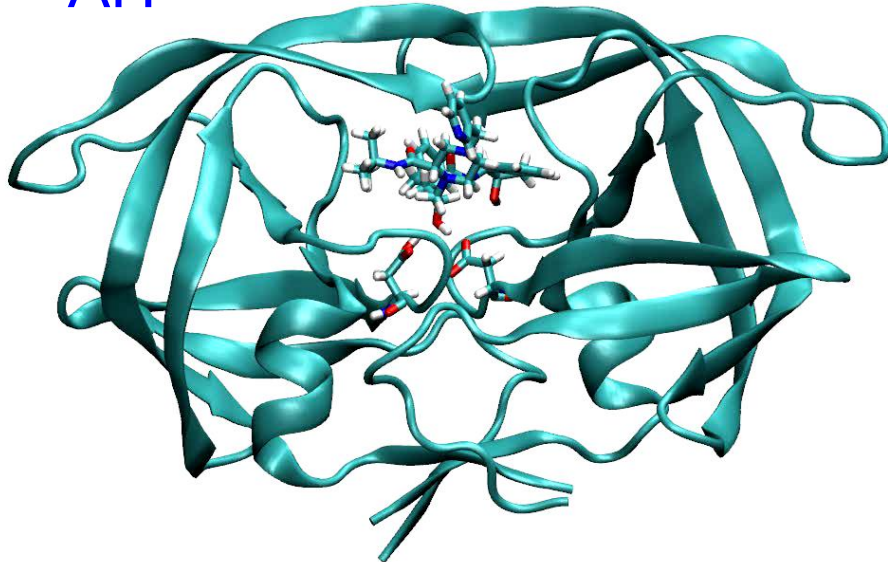
Intermediate

Water molecule
attached a lone pair
of amine

**New mechanism
for stabilization
of intermediate state**

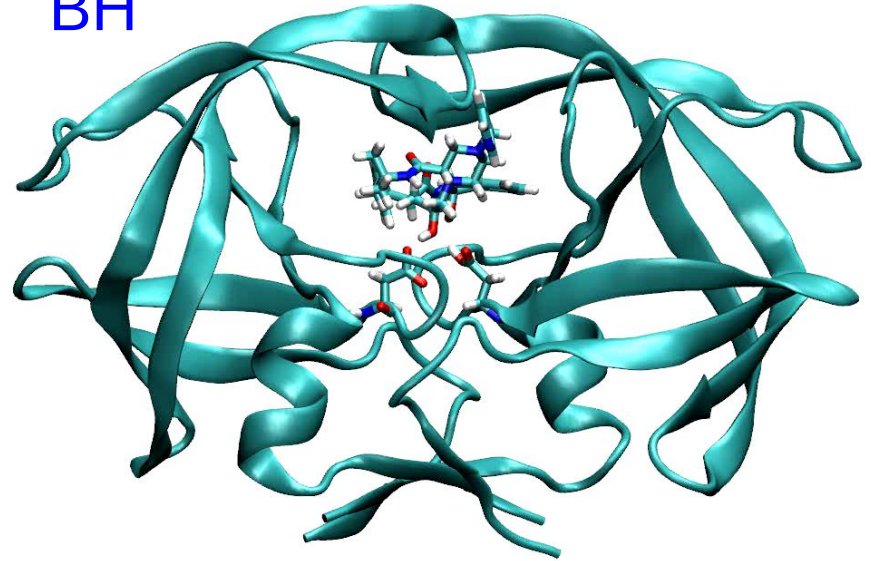
Inhibitor molecule (Indinavir) binding state (QM/MM)

AH



1100 ns

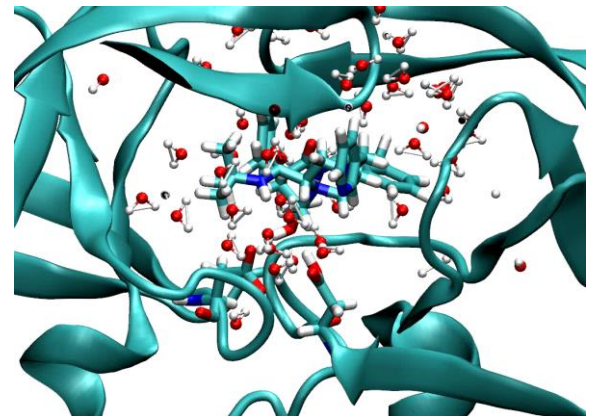
BH



700 ns

Two types of protonated
states(AH, BH)

- BH: water molecules entered between protease and inhibitor
- AH: more stable than BH



Summary

- Different protonated states caused different structural changes for both natural substrate and inhibitor cases
→ **Reasonable state is spontaneously determined**
- We clarified new stabilization mechanism of intermediate state
- We will calculate mutation case of protease
→ **Comparing it with wild type and try to make TS analog drug**